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# Investigation of bioremediation of arsenic by bacteria isolated from contaminated soil

Hadis Ghodsi<sup>1</sup>, Mehran Hoodaji<sup>1</sup>\*, Arezoo Tahmourespour<sup>2</sup> and Mohammad Mehdi Gheisari<sup>3</sup>

<sup>1</sup>Department of soil science, Khorasgan (Isfahan) branch, Islamic Azad University, Isfahan, Iran. <sup>2</sup>Department of microbiology, Khorasgan(Isfahan) branch, Islamic Azad University, Isfahan, Iran. <sup>3</sup>Department of basic science, Khorasgan (Isfahan) branch, Islamic Azad University, Isfahan, Iran.

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The aims of this study are isolating arsenite-resistant bacteria from arsenic contaminated soil and the investigation of arsenite bioremediation efficiency by the most resistant isolates. Isolation of arseniteresistant bacteria and the minimum inhibitory concentration (MIC) were conducted by spread plate method and the agar dilution method on PHG-II agar plates supplemented with sodium arsenite respectively. The results showed that, 69 and 25% of arsenite resistant isolates were geram positive and negative bacilli, respectively. Its maximum MIC was 128 mM/L, which is related to such bacteria as Bacillus macerans, Bacillus megaterimand Corynebacterium vitarumen. There is a significant difference (P< 0.01) between three isolates in arsenite removal potential and arsenite bioaccumulation. The maximum percentage of arsenite removal potential (92%) and arsenite bioaccumulation (36%) were related to B. macerans. The removal efficiency of arsenite for B. macerans, C. (vitaromen) and B. megaterim were 60, 43 and 38% after 48 h of growth, respectively, while after 144 h of Bacillus macerans, Corynebacterium (vitaromen) growth and 120 h of Bacillus megaterimgrowth were 92, 80 and 73% respectively. The results also were shown the highest percentage of arsenite in biomass (36%), arsenate from oxidation (27%) were related to B. macerans, B. megaterium and B. megaterium. These results express the probability of finding more arsenic accumulating bacteria from the contaminated soil environment and can be concluded that arsenic resistant and/or accumulating bacteria, such as Bacillus sp., are widespread in the polluted soils and are valuable candidates for bioremediation of arsenic contaminated ecosystems.

Key words: Arsenite, bacteria, bioremediation, MIC.

## INTRODUCTION

Human activities over the centuries has contaminated many areas of developing and developed countries (Evangelou et al., 2007). Soil contamination with heavy metals is one of the great problems of modern societies. Heavy metals periodically increase in the environment due to industrial activities and technology development. Increasing of these pollutants in the environment is considered as a serious threat to human and environmental health (Banaa Araghi et al., 2010). Unlike many organic contaminants disintegrated in the soil, heavy metals are kept in the soil storage and according to their nature a group of pollutants are of most interest because of their danger and of course plenty of stability in most environments (Garbisu and Alkorta, 2001). Arsenic has long been an important environmental pollutant and in long term has been as a health risk to humans and other living organisms. In the past, arsenic compounds have been widely used in pesticides, herbicides and soil disinfectors, thus in some soils was in high concentrations (Pais and Jons, 1997). Arsenic is highly toxic metal element that annually threatens the health of millions of people in the world (Chen and Shao, 2009). Inorganic arsenic forms are more dangerous than other forms for human health and in terms of classification fall in cancer-causing ingredient (Andrews,

<sup>\*</sup>Corresponding author. E.mail: m\_hoodaji@khuisf.ac.ir. Tel/ Fax: 00983115354045.

2001). In recent decades following increasing environmental pollution by heavy metals, scientists attracted to biological purification methods. In most cases of cleaning the contaminated ecosystems with chemical methods involves heavy costs and irreparable damages (Brooks, 1995; Nwuche and ugoji, 2008). Therefore one appropriate method is using biological method. Generally population and microbial activities in soil and water contaminated with the presence of metal will be reduced and modified (Kelly et al., 1998). On the other hand resistant microorganisms have evolved mechanisms to tolerate the toxicity of heavy metals. Application of microorganisms for heavy metals remediation is considered as a natural, stable and economical solution. Previous researches have described the isolation and characterization of arsenic resistant bacteria from different environments and have indicated that these bacteria are able to grow chemolithotrophically with oxvgen as an electron acceptor and As(III) as an electron donor (Duquesne et al., 2008; Santini et al., 2000). Arsenic-resistant bacteria play an important role in controlling the speciation and cycling of arsenic in the ecosystems (Inskeep et al., 2007). The aims of this study are isolating arsenite-resistant bacteria from arsenic contaminated soil and the investigation of arsenite bioremediation efficiency by high resistant isolates.

#### MATERIALS AND METHODS

#### Sampling

Soil samples were collected from the three different points of soil surface (0-20 cm) of the Research Farm (in Lavark, Najaf Abad) located in southwest of Isfahan, Iran. This soil was contaminated previously with arsenic through using of urban sewage sludge. The samples were mixed, transferred to the laboratory, passed through 2 mm sieve and used for physical-chemical and microbial analysis (Nwuche and ugoji, 2008).

#### Arsenic measurement

In this study to measure arsenic the spectrophotometry method was used along with a reagent called Leuco malachite green (LMG). In this method arsenic reacts with Potassium iodate (KIO<sub>3</sub>) in the acidic environment and iodine will be released. Released iodine oxidizes LMG to MG and changes the color to the color of malachite green. Detection range of arsenic concentration in this method is 0.09-0.9 micro g/ml. The MG dye shows maximum absorption at 617 nm (Revanasiddappa et al., 2007).

#### Arsenite measurement (As III)

Initially for the preparation of arsenite stock solution (1,000  $\mu$ g ml), amount of 0.1734g NaAsO<sub>2</sub> (sodium arsenite) resolved in 100 ml deionized distilled water. The standard solutions including 0.9-9.0  $\mu$ g of arsenite removed and poured in 10 ml volumetric balloons. Then 1 ml of Potassium iodate 1%, 0.5 ml of 1 M hydrochloric acid were added and the reaction mixture was shaked for 2 min. Then 0.5 ml of 0.05 LMG was added with shaking. Finally 2 ml acetat buffer (pH =4.5) was added and heated (40°C) in a water bath for 5 min, cooled and diluted with distilled water. After 5 min, absorbance of the dye was measured at 617 nm against the reagent blank. The concentration of arsenic (III) content was established by reference to the calibration graph (Revanasiddappa et al., 2007).

#### Arsenate measurement (As V)

After filtering of this sample a certain volume will be removed, then 0.5 ml of 5% Kl and 5 M HCL were added to the samples. All of the available arsenate were reduced to arsenite. In order to remove yellow to brown color which is due to the high amount of released iodine, the droplets of ascorbic acid were added (Pillai et al., 2000) then the amount of total arsenic in samples were measured by the method of arsenite measurment.

#### Soil arsenic measurement

One gram of soil sample was placed in the nickel plate that already covered its bottom with NaOH. Heat the Nickel plate in order to NaOH be fully melted and mixed with soil (alkaline digestion). After cooling, the nickel plate was immersed in HCL (0.5 N) and waited in order to be digested slowly (Almond, 1953) then the amount of arsenite was measured.

#### Isolation of arsenite-resistant bacteria

One gram of each soil samples was used to provide series. 0.1 ml of each dilution was added to each of PHG-II agar plates (4 g pepton, 1 g yeast extract and 2 g glucose and 15 g agar per liter) supplemented with 0.5 mM sodium arsenite (pH=7) by spread-plate method. The plates were incubated at 30°C for 3-5 days. This experiment was conducted in three replicates. After isolation of resistant colonies, their enrichment, purification and identification were done by the help of Gram Staining and biochemical tests (catalase- Licetinase- citrate – MR- VP -manitol fermentation and acid production - fermentation of glucose, sucrose and galactose - nitrate reduction test, urease and Esculine Hydrolysis) (Sneat et al., 1989; Cappuccino and Sherman, 1996).

#### Minimum inhibitory concentration (MIC) determination

The (MIC) of arsenite at which no colony growth occurred was determined by the agar dilution method. PHG-II agar plates supplemented with different concentration of arsenite the level of resistance (0.5, 1, 2, 4, ... and 192 mM /L) were inoculatedas aseptically with a culture of bacterial isolates in exponential growth phase. The plates were incubated for 48 h at 35°C. Minimum concentration of arsenite allowing growth of the isolates was an indication of positive tolerance (Hassen et al., 1998).

#### Growth curve and arsenite removal by bacterial strains

The growth curve of the most resistant bacteria at sub MIC concentration of arsenite were monitored by measuring the optical density (OD) of the cultures at 600 nm using a spectrophotometer. At each intervales a certain volum of medium was removed, after measuring OD at 600 nm. It was centrifuged and filtered. Than, the arsenite and arsenate concentration were measured by the above mentioned method (Chen and Shao, 2008; Revanasiddappa et al., 2007; Pillai et al., 2000). At the end of growth phase arsenite concentration in bacterial cell structure was measured by the method of Takeuchi et al., 2007. Briefly, the culture media were shake at 100 rpm, centrifuged at 5000 ×g at 4°C for 20 min. The

Table 1. Soil chemical and physical properties.

Depth		Clay	Silt	Sand	mLl	EC	$\mathbf{P}_{ava}$	K <sub>ava</sub>	<b>As</b> <sub>Total</sub>	As <sup>5+</sup>	As <sup>3+</sup>
(cm)	Texture		(%)		рп	(dS.m <sup>-1</sup> )			(mg.kg ์	1)	
0-20	SiCL	35.83	48.83	15.34	7.63	2.6	10	203	60.2	40.3	19.7

Table 2. The bichemical tests and MIC of the greatest arsenite resistant bacteria.

Bacteria test	Strain 4	Strain 8	Strain 10	
Spore	+	-	+	
Catalase	+	+	+	
Manitol	+	ND	+	
VP	-	-	-	
MR	+	+	+	
Nitrate	+	+	+	
Citrate	-	-	+	
Glucose	+	+	+	
Sucrose	ND	+	+	
Urea	ND	+	ND	
Probable general and species	B. macerans	C. (vitaromen)	B. (megaterim)	
MIC	128	128	128	

pelletes were washed twice with distilled water and placed in an oven with 100°C temperature for drying. The dried sampels were weighted and digested with nitric asid. Then the arsenite concentration was measured at 617 nm by the spectrophotometric method.

#### Statistical analysis

Statistical analysis was conducted using the SPSS System software. For comparison of means the Duncan test was used at the 5% probability level.

### RESULTS

#### Soil physical and chemical properties

The physical and chemical properties of soil is presented in Table 1.

#### **Resistance to arsenite**

The results showed that, 69 and 25% of arsenite resistant isolates were geram positive and negative bacilli, respectively and 6% of them were gram positive cocci.

According to MIC determination results, the greatest resistance to arsenite has been related to gram positive bacilli. Its maximum MIC and MBC were 128 and 192 mM/L respectively, which is related to such bacteria as *B. macerans, B. megaterim* and *C. vitarumen* (Table 2).

# Growth curve and arsenite removal by bacterial strains

In Figure 1 growth curves of *B. macerans, C. vitaromen, B. megaterium* were shown in 128 mM arsenite.

The removal efficiency of arsenite for *Bacillus* macerans, *Corynebacterium* (vitaromen) and *Bacillus* megaterimwere 60, 43 and 38% after 48 h of growth, respectively. While after 144 h of *Bacillus* macerans, *Corynebacterium* (vitaromen) growth and 120 h of *Bacillus* megaterim growth, the removal efficiency of arsenite were 92, 80 and 73% respectively (Figures 2, 3 and 4).

Finally, the percentage of arsenite in bacterial mass (bioaccumolation), arsenate from oxidation and remained arsenite were determined. The results were shown in Figure 5. The highest percentage of arsenite in biomass (36%), arsenate from oxidation (27%) were related to *B. macerans, B. megaterium* and *B. megaterium*.

There is a significant difference (P< 0.01) between three isolates in arsenite removal potential and arsenite bioaccumulation (Figure 6). The maximum percentage of arsenite removal potential (92%) and arsenite bioaccumulation (36%) were related to *B. macerans*.

#### DISCUSSION

The first step in the identification of bacteria with the ability of bioremediation is isolation of resistant bacteria



Figure 1. Bacterial growth curves of *Bacillus macerans, Corynebacterium vitaromen and Bacillus megaterium* in 128 mM of arsenite.



Figure 2. Growth and arsenite removal curvs of Bacillus macerans.



Figure 3. Growth and arsenite removal curvs of Corynebacterium vitaromen.

which tolerate high concentrations of heavy metals (Trevors et al., 1985). Most arsenic resistant bacteria are separated from arsenic-rich environments. In natural environments, the number of arsenite resistant bacteria is less than arsenate resistant bacteria. Arsenate is more toxic than arsenite (Jackson et al., 2005). Among the isolated resistant strains from contaminated soil, three strains demonstrated dramatic resistance to arsenite 128



Figure 4. Growth and arsenite removal curvs of Bacillus megaterium.



Figure 5. Arsenite in biomass (bioaccumulation), arsenate from oxidation and remaind arsenite percentage of A: Bacillus macerans, B: Corynebacterium vitaromen and C: Bacillus megaterium.



Figure 6. The comparision of arsenite removal potential in 3 isolates.

Mm. These arsenite-resistant strains were probably *B. macerans*, *C. vitaromen* and *B. megaterium*. Concentrations of metals used in this study are also used in the similar studies for bacteria that their medium contains extracted yeast. Abu-shnab et al. (2003) showed that in a contaminated soil the 11.1% of isolated bacteria

were resistant to As with the MIC of 20 mM/L. High levels of soil metal concentration can lead to achieving such a high MIC in resistant strains. Also Chitpirom et al. (2009), in Thailand, isolated arsenic-resistant bacteria from tannery effluent and agricultural soils that were belonged to *Klebsiella*, *Pseudomonas*, *Comamonas* and Enterobacter genera with the MIC of 40 mM (arsenite) and 400 mM (arsenate). Pepi et al. (2007) isolated 3 arsenic resistant genera (Aeromonas, Bacillus and Pseudomonas) from contaminated sediments with the MIC of 16.66 mM (arsenite) and 133.47 mM (arsenate). They also concluded that these bacteria are suitable for arsenic bioremediation in contaminated sediments. In a study by Luis et al. (2006) in Spain with the aim of biological removing of arsenic, Corvnebacterium glutamicum with over 60 mM arsenite resistance identified as one of the most tolerant species to arsenic. This results are in agreement with our findings but our isolates could tolerate the higher concentration of arsenite that was related to high level of arsenite in soil. In the study after 144 h of B. macerans, C. vitaromen growth and 120 h of Bacillus (megaterium) growth, the removal efficiency of arsenite were 92, 80 and 73% respectively. The highest percentage of arsenite in biomass (36%), arsenate from oxidation (27%) were related to B. macerans, B. megaterium and В. megaterium. Among resistant isolates, B. macerans was able to remove 92% of arsenite in the medium and also store 36% of it in the cell mass which is introduced as superior strain in this regard. Studies by Mondal et al. (2008) on three strains of Ralstonia eutropha. Pseudomonas putida and Bacillus indicus showed that these strains were able to remove (67, 60 and 61% respectively) arsenic from wastewater.

Takeuchi et al. (2007) could isolate a non- genetically engineered potent arsenic accumulating bacterium, Marinomonas communis, from marine and non marine environment in Japan which accumulated 2290 µg Asg dw<sup>-1</sup> of arsenic in presence of 5 mg As/ I of arsenat (45.8%). Our results are in agree with Takeuchi et al. (2007) and although details of such mechanisms are not yet clear, accumulation of arsenic into the cell would be a result of higher uptake and lower efflux. The high effective concentration of As in this study and previous study (Takeuchi et al., 2007) could be related to the arsenic resistance systems such as presence of regulatory protein of the ars operon that has a specific binding site available for arsenite. However, presence of ars operon in bacteria is known to extrude arsenate from the cell by an efflux system. Consequently, arsenic is not accumulated in bacteria. Furthermore, the other known arsenic-resistant system, the phosphate-specific transport (Pst) system, would also lead to lower uptake of arsenat by the cell. Therefore, the present results obtained in our isolates and previousley isolated M. communis with its higher resistance and higher accumulation of arsenic contradict the known arsenicresistant systems, suggesting existence of an as yet unknown arsenic resistance system for these strains. Cai et al. (1998) also could isolate Pseudomonas strains without the ars operon with a yet unknown arsenic resistance system.

These results express the probability of finding more

arsenic accumulation bacteria from the contaminated soil environment. It can be concluded that arsenic resistant and/or accumulating bacteria are widespread in the polluted soil environment, and that arsenic-accumulating bacteria such as *Bacillus* sp. are valuable candidates for arsenic contaminated ecosystems bioremediation.

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