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Compositional features of precambrian pegmatites of Ago-Iwoye area South Western, Nigeria

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The increase in global demand for rare metal Ta-Nb deposits has caused a resurgence of interest in the search for economically viable deposits. Precambrian pegmatites occurring as near vertical dykes have been studied, with the aim of determining their compositional features and possible economic values. Four thin sections were prepared from the pegmatites of the study area for petrographic study. A total of twenty five samples comprising whole rock pegmatites and extracts of mica were analyzed for major and trace elements using inductively coupled plasma-atomic emission spectrometry analytical technique (ICP-AES). Petrographic study shows that quartz, microcline, plagioclase are the main mineral constituents under transmitted light. From the results, the whole rock pegmatite is considerably siliceous, but with noticeable depletion of silica in the mica extract. Average Fe_2O_3 , Mgo, Mno, and TiO_2 values are low in all samples (< 2.00%). The samples are fairly enriched in Rb, Sr, Zr, but comparatively, poor in the rare metals Ta, Nb, W, Cs and Sn. Rare metal mineralization enrichment indices mainly, Ta vs Nb, Ta vs K/Cs, plots, show its depletion in rare metal mineralization, suggesting Ago-lwoye pegmatites to be barren in rare-metal mineralization when compared with other rare-metal pegmatites across the world.

Key words: Muscovite, pegmatite, mineralization, precambrian, amphibolites.

INTRODUCTION

The Ago-Iwoye study area lies between longitude 3° 54 E to 4° 00 E and latitude 6° 55 N to 7° 00 N Ijebu-Ode Sheet 280 N.E. (Figure 1). The area is also well accessible by networks of major and minor roads as well as untarred roads, with foot paths linking the environs together. In recent times there has been renewed interest in the study of pegmatites globally because of its attractive economic potentials. In Nigeria also, the study of pegmatites had arouse interest over the years for instance Jacobs and Webb (1946) studied the pegmatites of Nigeria and highlighted that it is restricted within a confine of 400 km NE-SE trending belt; however, studies carried out by Garba (2003) and OkunIola (2005) showed that the pegmatites are not restricted only to these confines. The occurrences in the Southeastern part of the

Nigeria, notably around Obudu hills were presumed to extend into Northeast Brazil (Garba, 2003; Ekwueme, 2004). The Nigerian pegmatites evolved during the time span of 600±530 Ma, (Matheis and Caen, 1983), which indicates the formation (Orogeny) in the periods of Pan African magmatism.

Precambrian pegmatites of Nigeria are known to host a variety of rare metals, namely tantalum, niobium, tin, tungsten, columbite as well as lithium and their various uses which include the production of microchips and microprocessors for computers and electronics, aircraft construction, casting, galvanizing, production of containers, metal wears. More importantly, the tantalum and niobium contained in this specialty metals are used for heat and corrosion resistant steels and alloys applied in space ships and gas turbines (Okunlola, 1998; Adekoya et al., 2003; Garba, 2003; Okunlola and Ogedengbe, 2003; Akintola, 2004; Okunlola, 2005; Okunlola and Jimba, 2006; Okunlola and Somorin, 2005;

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Figure 1. The location map Ago-lwoye study area showing the sampling points.

Okunlola and Akintola, 2007; Okunlola and Akintola, 2008). Okunlola (2005) classified the metallogeny of the

rare metal Ta-Nb pegmatites of Nigeria, outlining 7 broad fields namely Kabba - Isanlu, Ijero - Aramoko, Keffi-



Figure 2. Map showing Pan African province of basement complex. Adapted after Elueze (1985).

Nasarrawa, Lema -Ndeji, Oke Ogun, Ibadan -Osogbo and Kushaka - B/Gwari. The Ago-Iwoye Pegmatites occurrence which is parts of the Precambrian pegmatites in parts of Southwestern Nigeria is to be studied with the aim of elucidating their petrographic and geochemical features with a view to understanding their genesis and economic potentials.

METHODOLOGY

Systematic geological mapping followed by thin section petrographic studies of fresh whole rock samples was carried out. The whole rock and muscovite extracts of the pegmatite samples were then analyzed for major, minor, trace; and rare earth elements using inductively-coupled plasma atomic emission spectrophotometry (ICP-AES), at Activation Laboratories LTD. (ACTLABS) Ancaster, Ontario Canada. The geochemical analytical procedure involves addition of 5 ml of perchloric acid (HCIO₄), trioxonitrate (V) HNO₃ and 15 ml Hydrofluoric acid (HF) to 0.5 g of sample.

The solution was stirred properly and allowed to evaporate to dryness after it was warmed at a low temperature for some hours. 4 ml of hydrochloric acid (HCI) was then added to the cooled solution and warmed to dissolve the salts. The solution was cooled; and then diluted to 50 ml with distilled water. The solution is then introduced into the ICP torch as aqueous - aerosol. The emitted light by the ions in the ICP was converted to an electrical signal by a photo multiplier in the spectrometer, the intensity of the electrical

signal produced by emitted light from the ions were compared to a standard (a previously measured intensity of a known concentration of the elements) and the concentration then computed.

Geological setting, field description and petrography

Rocks of the precambrian basement complex underlie the project area. The precambrian basement of Africa can be divided into three large masses or cratons (Figure 2). These are the Kalahari craton, Congo and West African cratons. They are separated from each other by a number of mobile belts active in late precambrian and early Paleozoic times. The Nigerian Basement Complex lies east of the Congo craton in a mobile belt affected by the Pan African Orogeny Nigeria is underlain by precambrian basement complex rocks, younger granites of Jurassic age and Cretaceous to Recent sediments (Figure 3). The basement rocks occupy about half of the land mass of the country, and is a part of the Pan-African mobile belt lying between the West African and Congo cratons (Black, 1980). There are however contrasting documentation of the evolution of the basement rocks. However loosely, the basement is grouped into three major groups lithostratigraphically namely Migmatite-Gneiss Quartzite Complex: comprising biotite and biotite hornblende gneisses, guartzites and guartz schist. Schist Belts, comprising paraschists and meta igneous rocks, which include schist, amphibolites, amphibole schist, talcose rocks, epidotic rocks, marble and calc-silicate rocks. They are mainly N-S to NNE-SSW trending belts of low grade supracrustal (and minor volcanic) assemblages.

Other secondary rocks used in delineating them are carbonates, calc gneiss and banded iron formation (BIF) and older granites,



Figure 3. Generalized geological map of Nigeria (Oyawoye, 1972).

which include granite, granodiorite, diorite charnockite, pegmatites and aplites. The Ago-Iwoye study area covers the northeastern part of the map of liebu-Ode sheet 280 NE. The rock types observed in this area include coarse porphyritic granite, biotite- granite-gneiss, banded gneiss, quartzite and quartz schist, amphibolites schist; and pegmatite occurring as near vertical intrusions into the older rocks (Figure 4), these pegmatites trend in the NW-SE direction. The coarse porphyritic granite occurs as a massive body and mostly contains minerals like quartz, microcline feldspar, and muscovite. The biotite-granite-gneiss is a metamorphic rock in which mineral biotite is the most abundant. Other minerals include quartz, muscovite and microcline feldspar. These rocks underlie a substantial part of the study area and occur as massive outcrops. They are mainly porphyroblastic in texture and dominantly granitic in composition. Phenocrysts of feldspar are pervasive on many of the outcrops while some contain mafic xenoliths or relicts. The banded gneiss also underlies the study area; it is massive and consists of alternating bands of felsic minerals mainly plagioclase feldspars and quartz, and the dark bands consist of mafic mineral like biotite.

The quartzite outcrops which form prominent ridges occur as jointed rocks with conspicuous display of incipient schistose texture. Discordant intrusions of late quartz vein are common across the outcrops. The amphibolites on the other hand occur as low lying discontinuous lenses usually along the contact of the gneiss, they are fissile. The main minerals include quartz, plagioclase, microcline, biotite, and hornblende, while accessory minerals include chlorite, apatite and opaque minerals. The pegmatite occurs

as coarse in equigranular veins, milky white in appearance; the main mineral assemblages include quartz, microcline and muscovite with subordinate garnet. The petrography of Ago-Iwoye pegmatite reveals that the predominant mineralogical constituents include plagioclase, microcline, microperthite, quartz, biotite and accessory opaque minerals mainly schroll (Table 1). Quartz exhibits euhedral shape with wavy extinction. The plagioclase feldspar exhibits polysynthetic twinning with microperthite development which is exsolution growth while microcline displays cross-hatch twinning or pericline twinning and is often graphically intergrown with quartz; biotite generally occurs as dark brown platy grains within the samples (Figure 5a(i), b(i), c(i), d(ii), b(ii), c(ii), d(ii)).

RESULTS AND INTERPRETATION

The analytical results are presented in Tables 2 and 4. Major element distribution shows that the barren pegmatites of Ago-Iwoye are siliceous; this is not unrelated to the fact that pegmatite bodies, barren of rare – metal mineralization are ubiquitous in the Nigerian Pan-African basement. They are found associated with all the major lithologies of the basement, that is, gneisses, migmatites, schist and granitoids. The morphology and major mineral composition (quartz-feldspar-mica) are



Figure 4. Geological map of Ago-Iwoye area southwestern Nigeria.

mostly not different from those of the rare-metal types, which account for the reasons why the Ago-Iwoye pegmatites are being siliceous, despite the barren nature of the pegmatite of this study area. The silica content SiO_2 ranging between 61.23 and 81.13% with an average value of 73.74% for the whole rock samples of Ago-Iwoye pegmatites, while it also ranges between 48.18 and 57.47% with an average value of 52.65% for the mica extracts of this study area. These SiO_2 values for the

whole rocks and mica extract samples of Ago-Iwoye pegmatites are greater than 14% in all the samples and it is the major oxide used in classifying igneous rocks using the Total Alkali Silica (TAS) diagram which divides the rocks into ultra basic, basic, intermediate and acidic on the bases of their silica content (Le maitre et al., 1989; Cox et al., 1979). From this classification, it can be inferred that the Ago-Iwoye pegmatite was derived from igneous protolith, which is acidic. Further, the values

| Minerals | P5 (%) | P6 (%) | P7 (%) | P8 (%) |
|--------------------|--------|--------|--------|--------|
| Plagioclase (Pl) | - | - | 15 | 35 |
| Microperthite (Mp) | 60 | - | - | - |
| Quartz (Q) | 30 | 30 | 10 | - |
| Biotite (B) | - | - | 10 | - |
| Microcline (M) | - | 65 | 60 | - |
| Accessories (A) | 10 | 5 | 5 | 65 |

 Table 1. Average modal composition (%) of minerals in Ago-Iwoye pegmatites.

P5-P8 Represent photomicrographs of pegmatites from Ago-Iwoye study area.

Table 2. Major element oxide composition of Ago-Iwoye Pegmatites (weight %).

| Oxides | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 |
|--------------------------------|-------|------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| SiO ₂ | 71.72 | 72.45 | 73.84 | 61.23 | 76.98 | 71.66 | 70.54 | 71.93 | 67.14 | 72.2 | 74.61 | 73.68 | 74.23 | 73.59 | 78.26 |
| AI_2O_3 | 15.81 | 16.39 | 15.99 | 15.86 | 11.96 | 16.82 | 17.13 | 16.72 | 17.63 | 16.53 | 12.21 | 13.21 | 14.66 | 15.86 | 12.05 |
| Fe_2O_3 | 1.31 | 0.75 | 0.99 | 2.46 | 2.46 | 1.2 | 1.47 | 0.82 | 1.47 | 1.18 | 2.5 | 2.79 | 1.87 | 1.24 | 2.15 |
| MnO | 0.081 | 0.021 | 0.025 | 0.038 | 0.048 | 0.041 | 0.031 | 0.019 | 0.025 | 0.066 | 0.033 | 0.02 | 0.007 | 0.004 | 0.007 |
| MgO | 0.13 | 0.09 | 0.2 | 0.78 | 0.43 | 0.12 | 0.14 | 0.15 | 0.16 | 0.18 | 0.21 | 0.19 | 0.06 | 0.02 | 0.05 |
| CaO | 0.77 | 0.88 | 1.01 | 3.91 | 1.53 | 0.61 | 0.52 | 0.98 | 0.09 | 0.85 | 0.6 | 0.61 | 0.42 | 0.28 | 0.19 |
| Na ₂ O | 4.8 | 5.34 | 4.79 | 2.26 | 2.18 | 5.16 | 4.91 | 5.88 | 3.03 | 6.91 | 3.71 | 4.28 | 5.13 | 5.36 | 4.08 |
| K ₂ O | 3.4 | 3.5 | 3.41 | 8.48 | 4.12 | 2.98 | 3.63 | 1.9 | 9.65 | 1.09 | 4.76 | 3.39 | 3.27 | 3.4 | 3.14 |
| TiO ₂ | 0.041 | 0.03 | 0.032 | 0.377 | 0.141 | 0.04 | 0.037 | 0.036 | 0.017 | 0.019 | 0.195 | 0.241 | 0.079 | 0.102 | 0.112 |
| P_2O_5 | 0.1 | 0.07 | 0.06 | 2.62 | 0.12 | 0.14 | 0.06 | 0.1 | 0.16 | 0.11 | 0.03 | 0.07 | 0.03 | 0.04 | 0.03 |
| LOI | 0.84 | 0.6 | 0.66 | 0.78 | 0.66 | 1.07 | 1.22 | 0.99 | 0.37 | 0.68 | 0.03 | 0.13 | 0.52 | 0.88 | 0.16 |
| Total | 99 | 100.1 | 101 | 98.81 | 100.6 | 99.82 | 99.71 | 99.52 | 99.73 | 99.82 | 98.9 | 98.62 | 100.3 | 100.8 | 100.2 |
| | | | | | | | | | | | | | | | |
| Oxides | 16 | 1 | 7 | 18 | 19 | | 20 | 21 | 2 | 2 | 23 | | 24 | | 25 |
| SiO ₂ | 75.9 | 8 81 | .13 | 79.66 | 76.64 | 4 | 77.38 | 57.47 | 48. | 41 | 53.3 | 8 | 48.18 | | 55.8 |
| AI_2O_3 | 14.1 | 6 11 | .57 | 12.24 | 13.6 | 5 | 13.14 | 24.69 | 31 | .2 | 28.4 | Ļ | 30.91 | | 26.77 |
| Fe ₂ O ₃ | 1.2 | 0. | 89 | 1.1 | 1.18 | 1 | 0.75 | 3.82 | 3. | 6 | 3.15 | 5 | 4.08 | | 3.31 |
| MnO | 0.05 | 0 . | 02 | 0.04 | 0.08 | | 80.0 | 0.06 | 0.0 | 05 | 0.04 | Ļ | 0.04 | | 0.04 |
| MgO | 0.11 | 0. | 08 | 0.17 | 0.17 | | 0.06 | 0.57 | 0. | 5 | 0.69 |) | 0.59 | | 0.46 |
| CaO | 0.57 | 0 . | 42 | 0.32 | 0.62 | | 0.49 | 0.03 | <0. | .01 | 0.01 | | 0.02 | | 0.11 |
| Na ₂ O | 5.43 | 3. | 34 | 2.36 | 4.93 | | 5.22 | 0.61 | 0.4 | 49 | 0.54 | Ļ | 0.65 | | 0.47 |
| K ₂ O | 1.66 | i 1. | 69 | 2.79 | 2.08 | 1 | 2 | 8.08 | 9.9 | 98 | 8.85 | 5 | 9.95 | | 8.4 |
| TiO ₂ | 0.05 | 0 . | 04 | 0.03 | 0.04 | | 0.02 | 0.17 | 0.2 | 23 | 0.16 | 6 | 0.2 | | 0.26 |
| P_2O_5 | 0.06 | 0 . | 07 | 0.05 | 0.09 | l | 0.07 | <0.01 | <0. | .01 | <0.0 | 1 | 0.01 | ~ | <0.01 |
| LOI | 0.7 | 0 | .7 | 1.2 | 0.5 | | 0.7 | 4.5 | 5. | 5 | 4.7 | | 5.3 | | 4.3 |
| Total | 99.98 | 8 99 | .99 | 99.98 | 99.99 | 9 | 99.95 | 99.94 | 99. | 96 | 99.9 | 5 | 99.96 | ę | 99.95 |

The numbers 1, 2, 3,......25 represent sample numbers. 1-20: whole rock pegmatite samples from Ago-Iwoye. 21-25: Muscovite extracts from Ago-Iwoye pegmatite.

obtained for the silica contents of this study area are comparable to the Ipetu Ijesha barren pegmatites (Elueze, 1982).

 Fe_2O_3 (0.75 to 2.79%; 3.15 to 4.08%), MnO (0.004 to 0.081%; 0.04 to 0.06%), MgO (0.02 to 0.78%; 0.46 to 0.69%), CaO (0.09 to 3.91%; 0.01 to 0.11%), TiO₂ (0.02 to 0.377%; 0.16 to 0.26%), P_2O_5 (0.03 to 2.62%; 0.001 to 0.01%), values for the whole rock and muscovite extract

samples of Ago-Iwoye pegmatites respectively are generally low. Mean contents of major oxides, Al_2O_3 (14.68%; 28.39%), Na_2O (4.46%; 0.55%), and K_2O (3.52%; 9.05%) for the whole rock and muscovite extract samples of Ago-Iwoye pegmatites respectively compare favorably with the Ipetu Ijesha barren pegmatites and Kafin Maiyaki barren pegmatites (Elueze, 1982; Garba, 2003; Okunlola, 2005). Trace and rare earth element





Figure 5. a(i) Photomicrograph of Pegmatite in transmitted light showing MicroPerthite (MP), and Quartz (Q). b(i) Photomicrograph of Pegmatite in transmitted light showing Microcline (M), and Quartz (Q). c(i) Photomicrograph of Pegmatite in transmitted light showing Microcline (M), Plagioclase (PL), Biotite(B) and Quartz (Q). d(i) Photomicrograph of pegmatite in transmitted light showing Plagioclase (PL). a(ii), b(ii), c(ii) and d(ii) are modal distributions of estimated minerals in Ago-Iwoye pegmatites; (Bar scale =20 mm; Resolution: 150 dpi).

data (Tables 3, 5 and 6) show the values of the following trace element Rb (106.85, 504.74 ppm), Cs (4.57, 23.24 ppm), Nb (21.88, 170.8 ppm), Ta (6.54, 12.52 ppm), Be (4.5, 10.2 ppm), Th (2.98, 0.28 ppm), Hf (3.82, 0.54 ppm), Y(16.15, 2.04 ppm), Sn (3.9, 46.6 ppm), Sr (89.43,

3.26 ppm) for the whole rock and muscovite extract samples of Ago-Iwoye pegmatites respectively, has values that are significantly lower than the averages for the rare metal pegmatites of Ijero-Aramoko-Ara, Kushaka-Birni Gwari, Oke –Ogun, Isanlu-Egbe and

Table 3. Trace and Rare earth element data of Ago-Iwoye Pegmatites (ppm).

| Elements | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 |
|----------|------------|----------|---------|--------------|--------------|----------|--------------|---------|-----------|----------|--------------|----------|-----------|---------|--------------|
| Та | 14.1 | 7.4 | 8.7 | 4.6 | 33.8 | 13.5 | 5.8 | 3.9 | 5.4 | 1.2 | 5.6 | 4.5 | 4.6 | 3.8 | 2.3 |
| Cs | 15.5 | 5.8 | 6.8 | 4.6 | 2.6 | 9.3 | 7.1 | 4.2 | 8.1 | 1.5 | 0.6 | 0.6 | 0.5 | 0.5 | 0.5 |
| Rb | 106 | 92 | 107 | 247 | 88 | 97 | 96 | 71 | 251 | 24 | 112 | 111 | 121 | 122 | 105 |
| Sn | 1 | 1 | 3 | 2 | 2 | 1 | 1 | 2 | 1 | 1 | 4 | 17 | 8 | 8 | 1 |
| Nb | 31 | 22 | 27 | 13 | 44 | 27 | 17 | 17 | 7 | 4 | 28 | 33 | 23 | 31 | 8 |
| Sr | 15 | 26 | 27 | 925 | 455 | 27 | 21 | 32 | 16 | 15 | 57 | 34 | 19 | 24 | 9 |
| Y | 9 | 11 | 13 | 55 | 9 | 6 | 3 | 7 | 2 | 6 | 112 | 36 | 7 | 9 | 12 |
| Ва | 13 | 19 | 22 | 2703 | 3646 | 41 | 46 | 15 | 37 | 32 | 427 | 342 | 168 | 218 | 75 |
| Hf | 1.5 | 1.5 | 1.6 | 0.8 | 3 | 0.4 | 0.8 | 0.4 | 0.3 | 0.8 | 16 | 17.4 | 8 | 7.4 | 13.8 |
| Th | 2.9 | 3.1 | 3.7 | 4.5 | 4.9 | 0.8 | 0.8 | 0.7 | 1 | 0.4 | 8.5 | 10.5 | 6.7 | 4.5 | 3.2 |
| W | 1 | 1 | 2 | 1 | 1 | 1 | 1 | 2 | 1 | 1 | 1 | 1 | 1 | 1 | 4 |
| Be | 5 | 3 | 3 | 2 | 1 | 12 | 6 | 6 | 2 | 7 | 4 | 4 | 3 | 3 | 2 |
| Zr | 26 | 27 | 19 | 18 | 90 | 10 | 16 | 9 | 10 | 19 | 402 | 566 | 213 | 221 | 367 |
| Ga | 18 | 16 | 18 | 18 | 12 | 20 | 17 | 17 | 12 | 14 | 22 | 31 | 27 | 29 | 27 |
| Zn | 40 | 30 | 50 | 80 | 50 | 30 | 30 | 30 | 100 | 90 | 90 | 190 | 50 | 40 | 40 |
| U | 3 | 2.6 | 3.3 | 2 | 2.4 | 9.3 | 2.5 | 2.1 | 1.9 | 2.3 | 2 | 1.9 | 1.4 | 1.4 | 2 |
| Ti | 0.9 | 0.5 | 0.8 | 1.1 | 0.3 | 0.4 | 0.4 | 0.3 | 1.5 | 0.2 | 0.5 | 0.5 | 0.6 | 0.5 | 0.3 |
| Cu | 70 | 10 | 50 | 50 | 90 | 40 | 20 | 170 | 20 | 80 | 80 | 50 | 20 | 20 | 50 |
| Li | 14 | 7 | 7 | 6.8 | 5 | 4.2 | 14 | 10.4 | 3.1 | 1.7 | 1 | 1 | 1 | 1 | 1 |
| | | | | | _ | | _ | | | | | | | | |
| Elements | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 |
| Co | 1 | 1 | 1 | 4 | 3 | 28 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| V | 5 | 5 | 5 | 34 | 12 | 5 | 5 | 5 | 5 | 5 | 5 | 1 | 5 | 5 | 5 |
| | 20 | 20 E | 20 | 20 | 20 | 20 | 20 | 20 | 20 | 20 | 20 | 20 | 20 | 20 | 20 |
| SC Dh | 4 | ວ ວຬ | с 20 | 1 | ১ 12 | 4 | 3 12 | 10 | 1 | 1 | 2 16 | ∠ 11 | 10 | 0 | 1 |
| PD | 20 | 20 | 20 | 20 | 13 | 10 | 13 | 12 | 40 2 | 24 | 10 | 11 2 | 12 | 9 | <i>1</i> |
| | 22 | 2 Q | 2 1 | 2 37 3 | 26.0 | ے 1 0 | ∠ 22 | 26 | 2 10 | ∠ 13 | ∠ ∕/2.8 | 5 6/1 | 2 | ∠ 22 | 2 2/1 |
| | 3.2 8.2 | 0.0 Q | 4 | 37.3 88.0 | 20.9 42.6 | 1.5 | 2.2 1/1 1 | 2.0 | 1.3 | 2.6 | 42.0 07 / | 120 | 25.1 | 2.2 | 24.1 70.7 |
| Pr | 0.2 | 1 18 | 1 23 | 12.2 | +2.0 5.63 | 0.53 | 0.53 | 0.56 | 5 0.52 | 0.29 | 13 3 | 18.6 | 1.06 | 0.73 | 9.08 |
| Nd | 47 | 5.2 | 5.5 | 48.3 | 18.9 | 2.5 | 2.1 | 2.3 | 21 | 14 | 48.5 | 68.3 | 47 | 4 | 32.8 |
| Sm | 1.7 | 1.6 | 17 | 11.3 | 3.4 | 0.8 | 0.6 | 0.6 | 0.5 | 0.4 | 13 | 18.2 | 1.7 | 14 | 8.6 |
| Eu | 1.01 | 0.15 | 0.15 | 3.79 | 1.31 | 0.1 | 0.1 | 0.11 | 0.06 | 0.08 | 1.72 | 2.23 | 0.08 | 0.1 | 0.59 |
| Gd | 1.2 | 1.5 | 1.9 | 11.9 | 3 | 0.8 | 0.7 | 0.8 | 0.4 | 0.5 | 14.6 | 21.8 | 2.1 | 2.7 | 7.7 |
| Tb | 0.3 | 0.4 | 0.5 | 1.8 | 0.4 | 0.2 | 0.1 | 0.2 | 0.1 | 0.2 | 3.2 | 4.2 | 0.7 | 0.9 | 1.7 |
| Dy | 1.9 | 2.3 | 2.5 | 9.7 | 2.1 | 1.4 | 0.7 | 1.4 | 0.4 | 1 | 20.9 | 25.8 | 4.9 | 6.4 | 11.7 |
| Ho | 0.3 | 0.4 | 0.4 | 1.9 | 0.4 | 0.2 | 0.1 | 0.1 | 0.2 | 4 | 4 | 5.5 | 1.2 | 1.5 | 2.3 |
| Er | 0.8 | 1 | 1.3 | 5.3 | 1.1 | 0.6 | 0.4 | 0.8 | 0.1 | 0.5 | 12.3 | 18 | 4.1 | 5 | 7.1 |
| Tm | 0.15 | 0.17 | 0.22 | 0.67 | 0.16 | 0.11 | 0.05 | 0.16 | 0.05 | 0.09 | 1.68 | 2.67 | 0.67 | 0.77 | 1 |
| Yb | 1.3 | 1.3 | 1.6 | 3.6 | 1 | 0.9 | 0.5 | 1 | 0.2 | 0.8 | 10.3 | 15.6 | 4.2 | 4.8 | 6.2 |
| Lu | 0.18 | 0.19 | 0.24 | 0.5 | 0.16 | 0.13 | 0.07 | 0.15 | 0.04 | 0.12 | 1.34 | 2.21 | 0.63 | 0.69 | 0.84 |
| | | | | | | | | | | | | | | | |
| Elements | 16 | | 17 | 18 | 19 | 20 | | 21 | 22 | | 23 | | 24 | | 25 |
| Та | 2.2 | 2 | 1.8 | 5 | 1.5 | 1 | 1 | 0.9 | 11 | | 15.4 | | 11.6 | 1 | 3.7 |
| Cs | 7.4 | l _ | 4.9 | 3.9 | 3.6 | 3.4 | 3 | 39.9 | 24. | 7 | 14.6 | | 17.7 | 1 | 9.3 |
| Rb | 105 | .7 | 65.9 | 90.3 | 65.1 | 59.9 | 6 | 78.5 | 529 | .3 | 348.2 | | 544.2 | 3 | 93.3 |
| Sn | 7 | • | 3 | 4 | 3 | 8 | - | 63 | 15 | , | 18 | | 24 | | 113 |
| ND | 37. | 8 | 18.9 | 20.9 | 13.2 | 14.8 | 2 | 34.7 | 137 | .4 - | 119.7 | | 138.7 | 2 | 14.5 |
| Sr | 10. | 4 | 10 | 30.1 | 18.3 | 17.7 | | 1.1 | 1.5 |) | 5.7 | | 2.3 | | 5.7 |
| Y De | 5 | | 1.3 | 11.9 | 3.3 | 4.4 | | ు ఎఎ | 0.2 | <u> </u> | 3.5 | | 1.1 47 | | ∠.4 40 |
| ва | 31 | | 10 | 57 | 11 | 22 | | 32 | 25 |) | 95 | | 17 | | 40 |

| Table | 3. | Contd. |
|-------|----|--------|
|-------|----|--------|

| Hf | 0.4 | 0.1 | 0.7 | 0.9 | 0.6 | 0.7 | 0.4 | 0.4 | 0.5 | 0.7 |
|----------|------|------|------|------|------|------|------|------|------|------|
| Th | 0.9 | 0.2 | 1.1 | 0.7 | 0.5 | 0.2 | 0.2 | 0.6 | 0.2 | 0.2 |
| W | 1.4 | 2.6 | 3.8 | 1.7 | 1.6 | 11.4 | 19.3 | 22.9 | 22.3 | 17.1 |
| Be | 6 | 5 | 5 | 6 | 5 | 9 | 13 | 12 | 9 | 8 |
| Zr | 6.1 | 1.6 | 7 | 16.6 | 8.8 | 5.3 | 3.8 | 4.8 | 6.3 | 7.1 |
| Ga | 19.5 | 15.4 | 16 | 15.6 | 15.6 | 90.8 | 90 | 72.4 | 97.9 | 89.9 |
| Zn | 16 | 3 | 1 | 5 | 443 | 18 | 6 | 8 | 6 | 10 |
| U | 0.9 | 0.5 | 2.3 | 1.6 | 1.2 | 1.1 | 0.1 | 1 | 0.5 | 0.8 |
| Ti | 0.1 | 0.1 | 0.1 | 0.1 | 0.1 | 0.3 | 0.2 | 0.1 | 0.1 | 0.1 |
| Cu | 12.5 | 4 | 1.6 | 2.7 | 1.4 | 2.6 | 0.4 | 0.7 | 1.2 | 1.2 |
| Li | 3.4 | 0.2 | 0.7 | 0.4 | 5.8 | 16.1 | 13.6 | 11.3 | 14.2 | 0.2 |
| | | | | | | | | | | |
| Elements | 16 | 17 | 18 | 19 | 20 | 21 | 22 | 23 | 24 | 25 |
| Со | 2.1 | 0.6 | 0.8 | 1.2 | 1 | 1.8 | 0.5 | 1.2 | 0.6 | 1 |
| V | 8 | 8 | 8 | 8 | 8 | 8 | 18 | 8 | 8 | 8 |
| Ni | 3.9 | 2.5 | 1.3 | 1.7 | 3 | 1.6 | 0.6 | 1.1 | 0.6 | 1.8 |
| Sc | 8 | 4 | 3 | 2 | 3 | 53 | 29 | 24 | 21 | 42 |
| Pb | 19.4 | 5 | 3.1 | 14.1 | 14.4 | 1.2 | 0.8 | 1.3 | 0.6 | 4.7 |
| Мо | 0.8 | 0.9 | 0.4 | 0.6 | 1 | 0.7 | 0.5 | 0.7 | 0.5 | 0.7 |
| La | 2.2 | 0.4 | 2.3 | 1 | 2.5 | 0.1 | 0.1 | 0.9 | 0.1 | 0.7 |
| Ce | 3.8 | 0.7 | 4.7 | 2 | 2.9 | 1.5 | 0.3 | 2.5 | 0.1 | 1.2 |
| Pr | 0.49 | 0.09 | 0.59 | 0.22 | 0.51 | 0.07 | 0.02 | 0.29 | 0.02 | 0.27 |
| Nd | 1.4 | 0.3 | 2.6 | 0.6 | 1.7 | 0.6 | 0.3 | 1 | 0.3 | 1.2 |
| Sm | 0.54 | 0.09 | 0.83 | 0.15 | 0.45 | 0.2 | 0.09 | 0.33 | 0.07 | 0.31 |
| Eu | 0.07 | 0.04 | 0.15 | 0.05 | 0.09 | 0.03 | 0.02 | 0.05 | 0.02 | 0.06 |
| Gd | 0.69 | 0.16 | 1.23 | 0.22 | 0.53 | 0.3 | 0.05 | 0.38 | 0.07 | 0.34 |
| Tb | 0.17 | 0.04 | 0.32 | 0.07 | 0.12 | 0.07 | 0.01 | 0.08 | 0.02 | 0.07 |
| Dy | 0.97 | 0.21 | 1.96 | 0.41 | 0.73 | 0.53 | 0.08 | 0.51 | 0.15 | 0.37 |
| Ho | 0.17 | 0.03 | 0.36 | 0.09 | 0.15 | 0.1 | 0.02 | 0.11 | 0.03 | 0.06 |
| Er | 0.48 | 0.1 | 1.07 | 0.29 | 0.41 | 0.25 | 0.03 | 0.29 | 0.09 | 0.2 |
| Tm | 0.11 | 0.02 | 0.17 | 0.06 | 0.09 | 0.07 | 0.01 | 0.05 | 0.02 | 0.03 |
| Yb | 0.86 | 0.13 | 1.23 | 0.51 | 0.7 | 0.43 | 0.05 | 0.39 | 0.14 | 0.21 |
| Lu | 0.13 | 0.02 | 0.18 | 0.09 | 0.12 | 0.07 | 0.01 | 0.06 | 0.02 | 0.04 |

The numbers 1, 2, 3,......25 represent sample numbers. 1-20: Whole rock Pegmatite samples from Ago-Iwoye. 21-25: Muscovite extracts from Ago-Iwoye pegmatites.

Share but are comparable to the llesha barren pegmatites (Elueze, 1982; Okunlola, 2005) and the Nasarawa-Kafin-Maiyaki barren pegmatites (Garba, 2003). The Rb/Sr ratio is low when compared to other rare metal mineralized pegmatites of Nigeria (Matheis and Emofurieta, 1987; Okunlola and Ocan, 2002; Okunlola, 2005; Elueze, 1980, 1981) but compares with the barren Nasarawa pegmatites (Garba, 2003). The K/Rb ratio have values that are significantly higher than those of the rare metal pegmatites of Nigeria but are comparable with values of the barren pegmatites and granitoids (Kuster, 1990; Garba, 2003). From the plot of K/Rb versus Cs (Figure 6a), and the plot of K/Rb versus Rb (Figure 6c, d) (Staurove et al., 1969) for the mica extracts and whole rock pegmatites of Ago-Iwoye which separates barren fractionation sequence and mineralized

pegmatites show the Ago-Iwoye pegmatite samples as plotting in the barren field, confirming the low degree of fractionation.

From the works of Garba (2003), it is believed that extreme fractionation of lithophile elements such as Rb and Cs is a common geochemical feature of granitic pegmatites, especially the rare-metal bearing types. Late - stage progressive fractionation crystallization leads to decrease in K: Rb ratio suggesting metasomatism and invariably mineralization. However, this assertion is not in consonance with values obtained for the Ago- Iwoye pegmatites in which the K/Rb ratio was on the higher side due to no late-stage progressive fractionation crystallization; hence, no mineralization. Further, from the plot of K/Rb versus Cs (Figure 6a), and the plot of K/Rb versus Rb (Staurove et al., 1969). From Figure 6c and d,

| | Ago-Iwoye samples (pegmatite) | | | | | | | | |
|--------------------------------|-------------------------------|------------|---------------------------|------------|--|--|--|--|--|
| Elements | Whole rocl | k (N = 20) | Muscovite extracts(N = 5) | | | | | | |
| | Range | Average(%) | Range | Average(%) | | | | | |
| Si ₂ O | 61.23 - 81.13 | 73.74 | 48.18 - 57.47 | 52.65 | | | | | |
| Al ₂ O ₃ | 11.57 - 17.63 | 14.68 | 24.69 - 31.20 | 28.39 | | | | | |
| Fe ₂ O ₃ | 0.75 - 2.79 | 1.49 | 3.15 - 4.08 | 3.59 | | | | | |
| MnO | 0.004 - 0.081 | 0.04 | 0.04 - 0.06 | 0.05 | | | | | |
| Mgo | 0.02 - 0.78 | 0.18 | 0.46 - 0.69 | 0.56 | | | | | |
| CaO | 0.09 - 3.91 | 0.78 | 0.01 - 0.11 | 0.04 | | | | | |
| Na ₂ O | 2.26 - 6.91 | 4.46 | 0.47 - 0.65 | 0.55 | | | | | |
| K ₂ O | 1.09 - 9.65 | 3.52 | 8.08 - 9.98 | 9.05 | | | | | |
| TiO ₂ | 0.02 - 0.377 | 0.08 | 0.16 - 0.26 | 0.20 | | | | | |
| P ₂ O ₅ | 0.03 - 2.62 | 0.20 | 0.01 - 0.01 | 0.01 | | | | | |

 Table 4. Range and average values of major elements in the whole rock and muscovite extracts of Agolwoye pegmatites in mass fraction (weight %).

 Table 5. Range and averages of some of the trace elements in the whole rock and muscovite extracts of Ago-Iwoye pegmatites (ppm).

| | Ago-Iwoye samples (pegmatite) | | | | | | | | |
|----------|-------------------------------|--------------|--------------|-----------------|--|--|--|--|--|
| Elements | Whole ro | ock (N = 20) | Muscovite ex | ctracts (N = 5) | | | | | |
| | Range(ppm) | Average(ppm) | Range(ppm) | Average(ppm) | | | | | |
| Та | 1.0-33.8 | 6.54 | 10.9-15.4 | 12.52 | | | | | |
| Cs | 0.5-15.5 | 4.57 | 14.6-39.9 | 23.24 | | | | | |
| Rb | 24-251 | 106.85 | 378.2-678.5 | 504.74 | | | | | |
| Sn | 1-17 | 3.9 | 15-113 | 46.6 | | | | | |
| Nb | 4-44 | 21.88 | 138.7-243.7 | 170.8 | | | | | |
| Sr | 9-925 | 89.43 | 1.1-5.7 | 3.26 | | | | | |
| Y | 1.3-112 | 16.15 | 0.2-3.5 | 2.04 | | | | | |
| Ba | 10-3646 | 396.75 | 17-95 | 41.8 | | | | | |
| Hf | 0.3-17.4 | 3.82 | 0.4-0.7 | 0.54 | | | | | |
| Th | 0.2-10.5 | 2.98 | 0.2-0.6 | 0.28 | | | | | |
| W | 1-4 | 1.56 | 11.4-22.9 | 18.6 | | | | | |
| Be | 1-12 | 4.5 | 8-13 | 10.2 | | | | | |
| Zr | 1.6-566 | 102.66 | 3.8-71 | 5.46 | | | | | |
| Ga | 12-31 | 19.01 | 72.4-97.9 | 88.2 | | | | | |
| Zn | 1-443 | 70.4 | 6-18 | 9.6 | | | | | |
| Ti | 0.1-1.5 | 0.47 | 0.1-0.3 | 0.16 | | | | | |

It can be inferred that the Ago-Iwoye pegmatites are relatively unfractionated (*primitive*) when compared with the rare-metal pegmatites. While Rb is an indicator of the degree of fractionation in the granitic pegmatites, Cs appears to be the most important discriminator of the rare-metal pegmatites (Figure 6a). The degree of albitization is revealed by the triangular Ti-Sn-(Nb+Ta) discriminate plot, which is in the zone of non-albitization (Figure 7) for the Ago-Iwoye pegmatites. This plot also reveals a poor degree of albitization and it indicates a significant difference between the mineralized and samples (Matheis unmineralized pegmatite and Emofurieta, 1990; Okunlola and King, 2003; Okunlola and Somorin, 2005; De Kun, 1965; Jacobson and Webb, 1946). However, they are comparable to the rare metal barren pegmatites of Nasarawa, Kafin Maiyaki and Itakpe Central Nigeria (Garba, 2003; Okunlola, 2005). The variation diagram plot (Figure 6e) of Na₂O/Al₂O₃ versus K_2O/Al_2O_3 reveals the igneous ancestry of the pegmatite which plot in the granite-igneous field of Garrells and Mackenzie, thus indicating and suggesting a graniticigneous ancestry for the Ago-lwoye pegmatites. Samples also plot in the field of volcanic arc granites on the Rb versus Y+Nb diagram (Figure 6l), while crustal thickness during emplacement of these pegmatite bodies (Figure 6o) reached about 30 km as shown from the Rb/Sr plot



Figure 6. (a) Plot of K/Rb versus Cs for Ago-Iwoye pegmatite (Cerny, 1982); (b) Plot of Ta versus Ga for the Ago-Iwoye pegmatite; (c) K/Rb versus Rb distribution pattern in the Muscovite extracts of Ago-Iwoye Pegmatites. Arrow indicate normal differentiation trend (Staurove et al., 1969); (d) K/Rb versus Rb distribution pattern in the whole rock Pegmatite of Ago-Iwoye Study area. Arrow indicate normal differentiation trend after Staurove et al. (1969); (e) Plot of Na₂O/Al₂O₃ versus K₂O/Al₂O₃ (Wt%) showing variation diagram for the Field of Igneous and Meta Sedimentary Rocks of Ago-Iwoye Pegmatites (Garrels and Mackenzie, 1971). (f) Plot of Ta versus Cs for the pegmatites of Ago-Iwoye study area (Moller and Morteani, 1987); (g) Plot of Ta versus Rb for the pegmatites of Ago-Iwoye study area (Moller and Morteani, 1987); (g) Plot of Ta versus Rb for the pegmatites of Ago-Iwoye study area (Goaupp et al., 1984); (i) Plot of Ta versus Nb for the pegmatites of Ago-Iwoye study area; (j) Plot of Ta versus K/Cs ratio for the pegmatite's of Ago-Iwoye study area (Gordiyenko, 1971; Beus, 1966); (k) Plot of Ta/W ratio versus Cs for the pegmatites of Ago-Iwoye study area. The Ta/W ratio increases with increasing elements fractionation as indicated By Cs (Moller and Morteani, 1987); (l) Rb versus (Y+Nb) discriminant diagram for the whole rock sample pegmatites Of Ago-Iwoye =AOI compared to those of Olode = OD, Komu= IK, and Awo= AO, (Pearce et al., 1984). VAG-Volcanic Arc Granite, ORG- Oceanic Ridge Granite, WPG- Within Plate Granite, SCG- Syn-Collisional Granite; (m) Zr-Sio₂ Plots of the pegmatites of Ago-Iwoye compared to those of Olode=OD, Komu=Ik, and Awo=AO pegmatites of Ago-Iwoye pegmatites compared to those of Olode=OD, Komu=Ik, and Awo=AO pegmatites; (n) Sr-Rb Plots of the Ago-Iwoye pegmatites compared to those of Olode=OD, Komu=Ik, and Awo=AO pegmatites; (o) Plot of Rb-Sr for the pegmatites of Ago-Iwoye study area (Condie, 1976); (p) Plot of Ta/ (Ta+Nb) versus Mn/(Mn+Fe) variation plots of the Ago-Iwo

| Ratio | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 |
|--|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|
| K/Rb | 7.0666667 | 3.5384615 | 3.962963 | 0.267027 | 0.1934066 | 3.5925926 | 4.5714286 | 2.21875 | 15.6875 | 1.6 |
| K/Ba | 0.2170508 | 0.1528763 | 0.1286345 | 0.0026036 | 0.0009378 | 0.0603196 | 0.0654899 | 0.1051207 | 0.2164469 | 0.0282685 |
| Na/K | 1.2618955 | 1.3637485 | 1.2555737 | 0.2382175 | 0.4729556 | 1.5477275 | 1.2090268 | 2.7662077 | 0.2806573 | 5.6664703 |
| Rb/Sr | 0.0266194 | 0.0315723 | 0.0264482 | 0.0284921 | 0.0388544 | 0.0254959 | 0.0313806 | 0.0222086 | 0.0319065 | 0.0376913 |
| Ba/Rb | 0.1226415 | 0.2065217 | 0.2056075 | 10.94332 | 41.431818 | 0.4226804 | 0.4791667 | 0.2112676 | 0.1474104 | 1.3333333 |
| Zr/Hf | 17.333333 | 18 | 11.875 | 22.5 | 30 | 25 | 20 | 22.5 | 33.333333 | 23.75 |
| Sr/Rb | 0.1415094 | 0.2826087 | 0.2523364 | 3.7449393 | 5.1704545 | 0.2783505 | 0.21875 | 0.4507042 | 0.063745 | 0.625 |
| Rb/Cs | 6.8387097 | 15.862069 | 15.735294 | 53.695652 | 33.846154 | 10.430108 | 13.521127 | 16.904762 | 30.987654 | 16 |
| Ta/W | 14.1 | 7.4 | 4.35 | 4.6 | 33.8 | 13.5 | 5.8 | 1.95 | 5.4 | 1.2 |
| K/Cs | 0.1820426 | 0.5008017 | 0.4161704 | 1.5299026 | 1.3150723 | 0.2659249 | 0.424301 | 0.375431 | 0.988708 | 0.6030607 |
| Zr/Zn | 0.65 | 0.9 | 0.38 | 0.225 | 1.8 | 0.3333333 | 0.5333333 | 0.3 | 0.1 | 0.2111111 |
| Th/U | 0.2800655 | 0.5564464 | 1.0951854 | 6.7995671 | 0.7305957 | 0.7977748 | 0.7955643 | 1.2514365 | 9.8870802 | 2.8566032 |
| Ta/Nb | 0.4548387 | 0.3363636 | 0.3222222 | 0.3538462 | 0.7681818 | 0.5 | 0.3411765 | 0.2294118 | 0.7714286 | 0.3 |
| Nb/Ta | 2.1985816 | 2.972973 | 3.1034483 | 2.826087 | 1.3017751 | 2 | 2.9310345 | 4.3589744 | 1.2962963 | 3.3333333 |
| K ₂ O/Na ₂ O | 0.7083333 | 0.6554307 | 0.7118998 | 3.7522124 | 1.8899083 | 0.5775194 | 0.7393075 | 0.3231293 | 3.1848185 | 0.1577424 |
| Na ₂ O/Al ₂ O | 0.3036053 | 0.3258084 | 0.2995622 | 0.1424968 | 0.1822742 | 0.3067776 | 0.2866316 | 0.3516746 | 0.1718661 | 0.4180278 |
| K_2O/AI_2O_3 | 0.2150538 | 0.2135448 | 0.2132583 | 0.5346784 | 0.3444816 | 0.17717 | 0.2119089 | 0.1136364 | 0.5473625 | 0.0659407 |
| | | | | | | | | | | |
| Ratio | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 | 19 | 20 |
| K/Rb | 186.66667 | 185 | 242 | 244 | 210 | 14.283784 | 13.44898 | 23.153846 | 18.083333 | 17.617647 |
| K/Ba | 0.6966715 | 1.1285093 | 1.4022669 | 1.4091166 | 1.1614261 | 2.9238346 | 1.7665292 | 0.7560819 | 2.1185789 | 2.3329293 |
| Na/K | 1.9649123 | 3.2647059 | 6.3684211 | 5.0833333 | 11.666667 | 10.163462 | 6.59 | 3 | 3.557377 | 3.3841808 |
| Rb/Sr | 3.8125 | 3.0810811 | 1.3884298 | 1.7868852 | 0.7142857 | 0.2932829 | 0.1517451 | 0.6312292 | 0.1689708 | 0.3672788 |
| Ba/Rb | 25.125 | 32.528736 | 26.625 | 29.864865 | 26.594203 | 15.25 | 16 | 10 | 18.44444 | 14.666667 |
| Zr/Hf | 0.5089286 | 0.3063063 | 0.1570248 | 0.1967213 | 0.0857143 | 0.0983917 | 0.1517451 | 0.3333333 | 0.281106 | 0.2954925 |
| Sr/Rb | 0.0092513 | 0.0082262 | 0.0161534 | 0.0129434 | 0.0347451 | 0.0444398 | 0.1402531 | 0.0406214 | 0.1569265 | 0.0754455 |
| Rb/Cs | 5.6 | 4.5 | 4.6 | 3.8 | 0.575 | 1.5714286 | 0.6923077 | 1.3157895 | 0.8823529 | 0.625 |
| Ta/W | 6.5838733 | 4.688935 | 5.427546 | 5.64332 | 5.211772 | 0.1861668 | 0.2862308 | 0.5936977 | 0.4794978 | 0.4881765 |
| K/Cs | 4.4666667 | 2.9789474 | 4.26 | 5.525 | 9.175 | 0.38125 | 0.5333333 | 7 | 3.32 | 0.0198646 |
| Zr/Zn | 1.4740015 | 1.5740241 | 1.2740718 | 1.0214154 | 0.5680405 | 0.4883062 | 0.5366828 | 0.084814 | 0.144427 | 24.575247 |
| Th/U | 0.2 | 0.1363636 | 0.2 | 0.1225806 | 0.2875 | 0.0582011 | 0.0952381 | 0.2392344 | 0.1136364 | 0.0675676 |
| Ta/Nb | 5 | 7.3333333 | 5 | 8.1578947 | 3.4782609 | 17.181818 | 10.5 | 4.18 | 8.8 | 14.8 |
| Nb/Ta | 1.2830189 | 0.7920561 | 0.6374269 | 0.6343284 | 0.7696078 | 0.305709 | 0.505988 | 1.1822034 | 0.4219067 | 0.3831418 |
| K ₂ O/Na ₂ O | 0.3038493 | 0.323997 | 0.3499318 | 0.3379571 | 0.3385892 | 0.3834746 | 0.2886776 | 0.1928105 | 0.3611722 | 0.3972603 |
| Na ₂ O/Al ₂ O ₃ | 0.3898444 | 0.2566238 | 0.2230559 | 0.2143758 | 0.2605809 | 0.1172316 | 0.1460674 | 0.2279412 | 0.152381 | 0.152207 |

Table 6. Elemental ratios of selected major and trace elements of pegmatites from Ago Iwoye study area.

| Table | 6. | Contd. |
|-------|----|--------|
|-------|----|--------|

| Ratio | 21 | 22 | 23 | 24 | 25 |
|---|-----------|-----------|-----------|-----------|-----------|
| K/Rb | 616.81818 | 352.86667 | 61.087719 | 236.6087 | 69 |
| K/Ba | 0.2095498 | 0.3312961 | 0.0773117 | 0.4857356 | 0.174279 |
| Na/K | 0.0674807 | 0.0438861 | 0.0545396 | 0.0583917 | 0.0500126 |
| Rb/Sr | 0.009883 | 0.0156478 | 0.0210931 | 0.0151737 | 0.0177248 |
| Ba/Rb | 0.0471629 | 0.0472322 | 0.2728317 | 0.0312385 | 0.1017035 |
| Zr/Hf | 7.5714286 | 9.5 | 12 | 12.6 | 10.142857 |
| Sr/Rb | 0.0016212 | 0.0028339 | 0.0163699 | 0.0042264 | 0.0144928 |
| Rb/Cs | 17.005013 | 21.42915 | 23.849315 | 30.745763 | 20.378238 |
| Ta/W | 0.9561404 | 0.5699482 | 0.6724891 | 0.5201794 | 0.8011696 |
| K/Cs | 0.1680599 | 0.3353199 | 0.5030558 | 0.4665257 | 0.3612 |
| Zr/Zn | 0.2944444 | 0.6333333 | 0.6 | 1.05 | 0.71 |
| Th/U | 0.5707696 | 0.5294525 | 0.8384264 | 0.4443102 | 0.5087324 |
| Ta/Nb | 0.0464423 | 0.0800582 | 0.128655 | 0.0836337 | 0.0638695 |
| Nb/Ta | 21.53211 | 12.490909 | 7.7727273 | 11.956897 | 15.656934 |
| K ₂ O/Na ₂ O | 13.245902 | 20.367347 | 16.388889 | 15.307692 | 17.87234 |
| Na ₂ O/Al ₂ O | 0.0247064 | 0.0157051 | 0.0190141 | 0.0210288 | 0.017557 |
| K ₂ O/Al ₂ O ₃ | 0.327258 | 0.3198718 | 0.3116197 | 0.3219023 | 0.3137841 |

The numbers 1, 2, 325 represent sample numbers.1-20: Whole rock Pegmatite samples from Ago-Iwoye. 21-25: Muscovite extracts from Ago-Iwoye pegmatites.

of Condie (1976). The plots of Zr/SiO₂ (Figure 6m) and Sr/Rb (Figure 6n) reveal samples plotting completely in the magmatic *"M"* field suggesting pegmatite outcrops around Ago-Iwoye area to be genetically magmatic with no post magmatic metasomatism and poorly differentiated.

The Ta-Nb mineralization potential trend as shown from plot of Ta versus Ga (Figure 6b) shows the whole rock and muscovite extract samples of Ago-Iwoye pegmatites plotting far below the boundary of mineralization (Beus, 1966) line suggesting the pegmatites of this study area to be barren. In comparing with the global Ta-Nb endowments, especially from variation plots of Ta versus Cs (Figure 6f), Ta versus Nb (Figure 6h), Ta versus Rb (Figure 6g), Ta versus Cs + Rb (Figure 6i), Ta versus K/Cs (Figure 6j) and Ta/W versus Cs (Figure 6k), Ago-Iwoye pegmatites compare favorably with the barren pegmatites of Pullerstreuth (Moller and Morteani, 1987). They also compare favorably with the barren pegmatites and granitoids of Nasarawa and Kafin Maiyaki Northern Nigeria (Garba, 2003), Itakpe Area Central Nigeria (Okunlola and Somorin, 2005), and plot far below the highly endowed Tanco pegmatites. They also plot far below the marginally endowed White City, Cross Lake, Buckclaim, pegmatites all of Canada, Noumas South Africa (Moller and Morteani, 1987) and Sepeteri Nigeria (Okunlola and Akintola, 2007). These plots confirm the barren nature of the Ago-Iwoye Pegmatites. In the same vein they show the Ago-lwove samples plotting far below the boundary of mineralization lines of Beus (1966)

and Gordiyenko (1971), and also far below other rare metal pegmatites from across the world. Following after the classification criteria of pegmatites based on bulk chemistry signatures (Cerny, 1991), and the Ta/(Ta+Nb) versus Mn/(Mn+Fe) plot (Figure 6p). The Ago-Iwoye pegmatites are of the simple non-mineralized class. Following after the rare earth element fractionation trends of pegmatites as defined by Ce anomalies observed for the mineralized pegmatites of Northern Nigeria (Garba, 2003), Negative Ce anomalies of rare metal pegmatite are taken to be indicative of oxidizing conditions during mineralization and involve possible interaction between magmatic and melt fluids and host rocks over sometimes long distances (Graff, 1977: Piper, 1974: Curlet et al., 1974: Garba, 2003).



Figure 7. Triangular Ti-Sn-(Nb+Ta) plot for Ago-Iwoye pegmatite's (Kuster, 1990).

The absence of this trend in the samples of the Agolwoye pegmatites rules out the role of late stage or metasomatic fluids in the genesis of these rocks (Figure 8). This further confirms minimal or lack of metasomatic alterations or extensive post magmatic differentiation in their emplacement. This might also mean a close proximity of emplacement to their parent melt sources, which is *"in situ"*.

Conclusions

Near vertical dipping coarse grained pegmatite veins occur in association with coarse porphyritic granite, biotite-granite-gneiss, banded gneiss, amphibolites schist, quartzite and quartz schist around Ago-Iwoye area southwestern Nigeria. Petrographic determination shows they are enriched in Microcline and quartz and to a lesser extent in Plagioclase (albite), with interstitial muscovite and accessory biotite. Geochemical studies reveal that they are poorly fractionated. This shows nearness, to their parental melt sources or metamorphic progenitors. Rare metal indicative elements Ta, Nb, Rb, Cs and Sn are depleted in the rock unit while elemental ratio, K/Rb, Ba/Rb suggest low index of fractionation, poor fractionation and barren mineralization. Poor albitization is demonstrated in the Na/K ratio and the Plot of Ti-Sn-(Nb+Ta) after Kuster (1990). While Ta versus Cs, Ta versus Rb and Ta versus K/Cs confirms its apparently poor or barren mineralization compared to other pegmatites bodies in Nigeria and elsewhere in the world.

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Figure 8. Rare earth elements chondrite - normalized plots of Ago-Iwoye pegmatite.

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REFERENCES

- Adekoya JA, Kehinde-Phillips OO, Odukoya AM (2003). Geological distribution of Mineral resources in Southwestern Nigeria. Prospects for investment in mineral resources of Southwestern Nigeria, A.A. Elueze (ed), Nigerian Mining And Geosciences Society (NMGS); pp.15-56.
- Akintola AI (2004). Petrography and Petrogenetic features of rare metal pegmatites and associated amphibolitic schist of Sepeteri area, southwestern Nigeria. M.sc Thesis University of Ibadan (Unpublished), p. 74.
- Beus AA (1966). Distribution of tantalum and niobium in muscovites from granitic pegmatites. Geokhimiya, 10: 1216-1220.
- Black R (1980). Precambrian of West Africa. Episode, 4: 3-8.
- Cerny P (1982). Mineralogy of Rubidium and Cesium. In: Cerny, P. (ed.) Anatomy and classification of granitic pegmatite in Science and Industry. Mineralogical Association of Canada Short Course Handbook, 8: 1-39.
- Cerny P (1991). Geochemical and petrogenetic features of mineralization in rare element granitic pegmatites in the light of current research. Appl. Geochem., 18: 48-68.
- Condie KD (1976). Trace element geochemistry of Archean granite rocks from the Barberton Region, South Africa, AM. Mineral. Earth Planet Sc. Lett., pp. 300-400.
- Cox XG, MacDonald R, Hornug G (1979). Geochemical and Petrographic Province in The Karoo basalts of Southern Africa, AM. Mineral. Earth Planet. Sci. Lett., 24: 419-426.
- Curlet RL, Yeh L, Chau DS (1974). Rare earth elements in the Silurian Pelitic Schists from N.W Maine. Geochim. Cosmochim. Acta, 38: 289-400.
- DeKun N (1965). Mineral Resources of Africa. Elsevier Amsterdam, pp. 15-25.
- Ekwueme BN (2004). Pan-African Schist of Southeastern Nigeria and their relationship with Schists of Cameroon. Book of Abstracts 40th Nigerian Mining And Geosciences Society. International Conference, Maiduguri, p. 18.

- Elueze AA (1982). Geochemistry and petrogenetic setting of metasedimentary rocks of the schist belt of Ilesha area southwestern Nigeria. Precambrian research, pp. 303-318.
- Elueze AA (1980). Geochemical studies of proterozoic amphibolites and meta ultramafites in Nigeria schist belts; Implications for Precambrian crustal evolution. Unpublished PhD Thesis University of Ibadan.
- Elueze AA (1981). Dynamic metamorphism and oxidation of amphibolites of Tegina area Northwestern Nigeria. Prec. Res., 14: 368-379.
- Elueze AA (1985). Petrochemical and petrogenetic characteristics of Precambrian amphibolites of the Alawa district, north western Nigeria. J. Chem. Geol., 48: 29-41.
- Garba I (2003). Geochemical discrimination of newly discovered rare metal bearing and barren Pegmatites in the Pan-African (600 <u>+</u>150 Ma) basement of northern Nigeria. Applied Earth Science Transaction Institute Of Mining and Mettallurgy 13; 112: B287-B291.
- Garrels RM, Mackenzie FT (1971). Evolution of Igneous and Sedimentary Rocks. W.W. Norton and Company, Inc. New York, N.Y., p. 394.
- Gaupp R, Moller P, Morteani G (1984). Geology, Petrology and Geochemistry of Tantalum Pegmatites of untersuchungen. Monograph series, Min. Deposit 23: 124. Borntraeger Berlin Stuttgart.
- Gordiyenko VV (1971). Concentration of Li, Rb, and Cs in potash feldspar and muscovite as criteria for assessing rare metal mineralization in granite pegmatites. Int. Geol. Rev., 13: 134-142.
- Graff JL (1977). Rare Earth Elements as hydrothermal tracers during the formation of massive sulphide deposits in volcanic rocks. Econ. Geol., 72: 527-548.
- Jacobson RRE, Webbs JS (1946). The Pegmatite of Central Nigeria. Geol. Surv. Nig. Bull., 17: 40-61.
- Kuster D (1990). Rare metal pegmatites of Wamba, central Nigeria their formation in relationship to late Pan-African granites. Miner. Deposit., 25: 25-28.

Le Maitre RW, Bateman P, Dude, K.A.; Kelly, J.; Lameyre, L.E.; Bass, M.J.; Sabin, P.A.; Sehmid, R.; Sorensen, H.; Streckeisen, A. ; Noolley, A.R and Zaneltin, B. (1989). Classification of Igneous rock and glossary of terms. Blackwell Oxford, pp. 22-56.

Matheis G, Caen-Vachette (1983). Rb-Sr isotopic study of rare metal bearing and barren pegmatites in the Pan-African reactivation zone of Nigeria. J. Afr. Earth Sci., 1: 35-40.

- Matheis G, Emofurieta WO (1987). Nigerian Rare metal pegmatites and their lithological framework. J. Geol., 22: 271-291.
- Matheis G, Emofurieta WO (1990). The Older Tin province Rare-Metal Pegmatite in Nigeria. Technical University of Berlin (West). Special Research Project On Arid Areas And University Of Ife, Nigeria, p. 10.
- Moller P, Morteani G (1987). Geochemical exploration guide for Tantalum pegmatites. Econ. Geol., 42: 1885-1897.
- Okunlola OA (1998). Specialty metal potentials of Nigeria: Proceedings of 1st Mining in Nigeria Conference. Fed. Ministry of Solid Minerals, Abuja, Nigeria, pp. 67-72.
- Okunlola OA (2005). Metallogeny of Tantalite-Niobium Mineralization of Precambrian Pegmatites of Nigeria. Miner. Wealth, 104: 38-50.
- Okunlola, O.A. (2005). Metallogeny of Tantalite-Niobium Mineralization of Precambrian Pegmatites of Nigeria. Miner. Wealth, 104/2005, pp. 38-50.
- Okunlola OA, Akintola AI (2007). Geochemical features and rare metal Ta-Nb potentials Of Precambrian pegmatites of Sepeteri area, southwestern Nigeria. Ife. J. Sci., 9 (2): 203-214.
- Okunlola OA, Akintola AI (2008). Compositional features and rare metal (Ta-Nb) Potentials of Precambrian pegmatites of Lema-Ndeji area central Nigeria. Miner. Wealth, 149/2008, pp. 43-53.
- Okunlola OA, Jimba S (2006). Compositional trends in relation to Ta-Nb mineralization in Precambrian pegmatites of Aramoko-Ara-Ijero area, southwestern Nigeria. J. Mining Geol., 42(2): 113-126.
- Okunlola OA, King PA (2003). Process test work for the recovery of Tantalite-Columbite Concentrates from rare metal pegmatites of Nasarawa area, central Nigeria. Glob. J. Geo. Sci., 1(1): 85-103.

- Okunlola OA, Ogedengbe O (2003). Investment potentials of gemstone occurrences in Southwestern Nigeria In: Elueze, A.A (ed.) Prospects for Investment in Mineral Resources of Southwestern Nigerian, pp. 41-45.
- Okunlola OA, Somorin EB (2005). Compositional features of Precambrian pegmatites of Itakpe area, central Nigeria. Glob. J. Geol. Sci., 4(2): 221-230.
- Okunlola OA, Ocan OO (2002). The Expected Environmental Impact and Mitigation Studies of Organized Mining of Rare-Metal (Ta-Sn-Nb). Pegmatites around Keffi Area Northcentral Nigeria. J. Environ. Ext., 3: 64-68.
- Oyawoye MO (1972). The geology of the Nigerian Basement Complex. J. Nig. Min. Geol. Soc., 1: 87-102.
- Pearce JA, Harris NBW, Tindle AG (1984). Trace element discriminant diagrams for the tectonic interpretation of granitic rocks. J. Petrol., 25: 956-983
- Piper DZ (1974). Rare earth elements in sedimentary cycle: A summary. Chem. Geol., 14: 285-304.
- Staurove OD, Stolyarov LS, Isochewa EI (1969). Geochemistry and Origin of Verkh Iset Granitoid massif in central Ural. Geochem. Intern., 6: 1138-1146.